

**Amendments to the Claims:**

This listing of claims will replace all prior versions and listings of claims in the application:

1. (Currently Amended) A mesoporous material comprising spherical silica particles having a diameter of between 0.1 and 1.0 micron, said silica particles having an inner structure formed by chain stacks aligned substantially along the radius of the spherical particle, each of said chain stacks having pores interconnected with pores of an adjacent chain stack to form a nanotube structure in which pores of said nanotube structure extend in a substantially radial direction with respect to a center of said silica particles, said nanotube structure having a mean pore diameter between about 2.0 nm and about 4.0 nm, as measured by nitrogen adsorption, said spherical silica particles having a surface area between about 750 and about 1,050 m<sup>2</sup>/g and a mean pore volume between about 0.75 ml/g and about 1.0 ml/g.

2. (Original) The mesoporous material of claim 1, wherein said silica particles have the formula SiO<sub>2</sub>-M<sub>2</sub>O<sub>3</sub> where M is a trivalent metal.

3. (Original) The mesoporous material of claim 2, wherein M is selected from the group consisting of aluminum, gallium and lanthanide metals.

Claim 4 (Cancelled)

5. (Original) The mesoporous material of claim 1, wherein said pores have a mean diameter of about 3.5nm.

6. (Currently Amended) A mesoporous material comprising silica particles having a substantially elliptical shape with a major axis and a minor axis, said silica particles having an inner structure formed by chain stacks, each of said chain stacks having pores interconnected with pores of an adjacent chain stack to define a plurality of nanotubes, said nanotubes being oriented substantially parallel to said major axis and having a mean pore diameter of about 2.0 nm to about 4.0 nm.

7. (Original) The mesoporous material of claim 6, wherein said particles have a surface area of about 1200 m<sup>2</sup>/g to about 1561 m<sup>2</sup>/g.

8. (Original) The mesoporous material of claim 7, wherein said particles have a pore volume of about 0.75 ml/g.

9. (Original) The mesoporous material of claim 6, wherein said nanotubes are oriented unidirectional.

Claims 10-11 (Cancelled)

12. (Original) The mesoporous material of claim 6, wherein said silica particles have the formula SiO<sub>2</sub>-M<sub>2</sub>O<sub>3</sub> where M is a trivalent metal.

13. (Original) The mesoporous material of claim 12, wherein M is selected from the group consisting of aluminum and gallium.

14. (Original) The mesoporous material of claim 12, wherein M is a lanthanide metal.

15. (Currently Amended) A process for producing mesoporous silica particles comprising the steps of:

reacting an aqueous mixture of a surfactant, an organic cosolvent, an hydrolysis catalyst agent and a silica source to form a gel, wherein said surfactant and cosolvent are present in a molar ratio of about 0.001:1 to about 0.01:1; and

drying and calcining said gel to form mesoporous silica particles having an inner structure formed by chain stacks, each of said chain stacks having pores interconnected with pores of an adjacent chain stack to define a plurality of nanotubes, wherein said nanotubes have a mean pore diameter of about 2.0 nm to about 4.0 nm, said surfactant, organic cosolvent, hydrolysis catalyst agent and silica source being used in amounts effective to form spherical silica particles in which pores of said nanotubes extend in a substantially radial direction with respect to a center of said silica particles or to form elliptical silica particles having a major axis and a minor axis in which nanotubes are oriented substantially parallel to said major axis.

16. (Original) The process of claim 15, wherein said surfactant and cosolvent are present in a molar ratio of about 0.008:1 to about 0.006:1.

17. (Original) The process of claim 15, wherein said surfactant is selected from the group consisting of anionic, cationic and nonionic surfactants.

18. (Original) The process of claim 15, wherein said aqueous mixture includes an aqueous solution and where said surfactant is present in an amount of about 1.0 w% to about 8.0 wt% based on the weight of said aqueous solution.

19. (Original) The process of claim 15, wherein said cosolvent is selected from the group consisting of methanol, ethanol, propanol, butanol and acetone.

20. (Original) The process of claim 15, wherein said cosolvent is selected from the group consisting of a lower alkyl ketone, a lower alcohol, and mixtures thereof.

21. (Original) The process of claim 15, wherein said surfactant is cetyltrimethylammonium bromide and said hydrolysis catalyst is aqueous ammonium hydroxide.

22. (Original) The process of claim 15, wherein said surfactant and said hydrolysis catalyst are present in a molar ratio of about 0.07:1 to about 0.09:1.

23. (Original) The process of claim 15, wherein said cosolvent is a polar solvent having a surface tension and dielectric constant less than water, and where said polar solvent is present in an amount sufficient to lower the surface tension and dielectric constant of said aqueous mixture.

24. (Original) The process of claim 15, wherein said silica source comprises silica, an organosilicate, a silica metal oxide and mixtures thereof, where said silica metal oxide has the formula  $\text{SiO}_2\text{-MO}_x$  where M is a trivalent metal.

25. (Original) The process of claim 24, wherein said trivalent metal is selected from the group consisting of  $\text{Al}^{+3}$ ,  $\text{Ga}^{+3}$  and  $\text{La}^{+3}$ .

26. (Original) The process of claim 25, wherein said aqueous mixture further contains  $\text{Al}(\text{NO}_3)_3$  and where said silica particles have the formula  $\text{SiO}_2\text{-MO}_x$  where M is  $\text{Al}^{+3}$ .

27. (Original) The silica particles obtained by the process of claim 15, wherein said particles have a substantially spherical shape, a particle size of about 0.1 to about 1.0 microns and a surface area of about  $750 \text{ m}^2/\text{g}$  to about  $1050 \text{ m}^2/\text{g}$ .

28. (Original) The silica particles of claim 27, wherein said particles have a pore volume of about  $0.75 \text{ ml/g}$  to about  $1.0 \text{ ml/g}$ .

29. (Original) The silica particles obtained by the process of claim 15, wherein said particles have an elliptical shape with a particle size of about 0.1 microns to about 1.0 microns and a surface area of about  $1,000 \text{ m}^2/\text{g}$  to about  $1561 \text{ m}^2/\text{g}$ .

30. (Original) The silica particles of claim 29, wherein said pores of said nanotubes have a mean pore size of about 2.0 nm to about 4.0 nm.

31. (Original) The silica particles of claim 29, wherein said particles have a pore volume of about 0.75 ml/g.

32. (Original) The process of claim 15, wherein said aqueous mixture further comprises at least one compound selected from the group consisting of sodium aluminate, sodium gallate, a lanthanide metal nitrate and a lanthanide metal chloride and where said silica particles have the formula  $\text{SiO}_2\text{-MO}_x$  where M is  $\text{Al}^{+3}$ ,  $\text{Ga}^{+3}$ , or a trivalent lanthanide metal.

33. (Original) The process of claim 15, wherein said silica source is an organosilicate.

34. (Currently Amended) The process of claim ~~35~~ 33, wherein said organosilicate is tetraethylorthosilicate.

35. (Original) The process of claim 15, wherein said silica source sodium metasilicate.

36. (Currently Amended) The process of claim 15, wherein said cosolvent is selected from the group consisting of methanol, acetone and mixtures thereof, and where said mesoporous particles are ~~spherical~~ elliptical with pores extending ~~in a radial direction with respect to a center of said particles~~ substantially parallel to a major axis of said particles.

37. (Original) The process of claim 36, wherein said surfactant and cosolvent are included in said aqueous mixture in a molar ratio of about 0.009:1 to about 0.023:1.

38. (Original) The process of claim 36, wherein said aqueous mixture comprises said surfactant and said hydrolysis catalyst agent in a molar ratio of about 0.09:1 to about 0.07:1.

39. (Currently Amended) The process of claim 15, wherein said cosolvent is selected from the group consisting of ethanol, propanol and mixtures thereof, and where said mesoporous particles are spherical with pores in a substantially radial direction with respect to a center of said particles ~~elliptical shaped with pores are oriented in substantially the same direction.~~

40. (Original) The process of claim 38, wherein said surfactant and cosolvent are present in said aqueous mixture in a molar ratio of about 0.008:1 to about 0.006:1.

41. (Original) The process of claim 39, wherein said aqueous mixture comprises said surfactant and said hydrolysis catalyst in a molar ratio of about 0.09:1 to about 0.07:1.